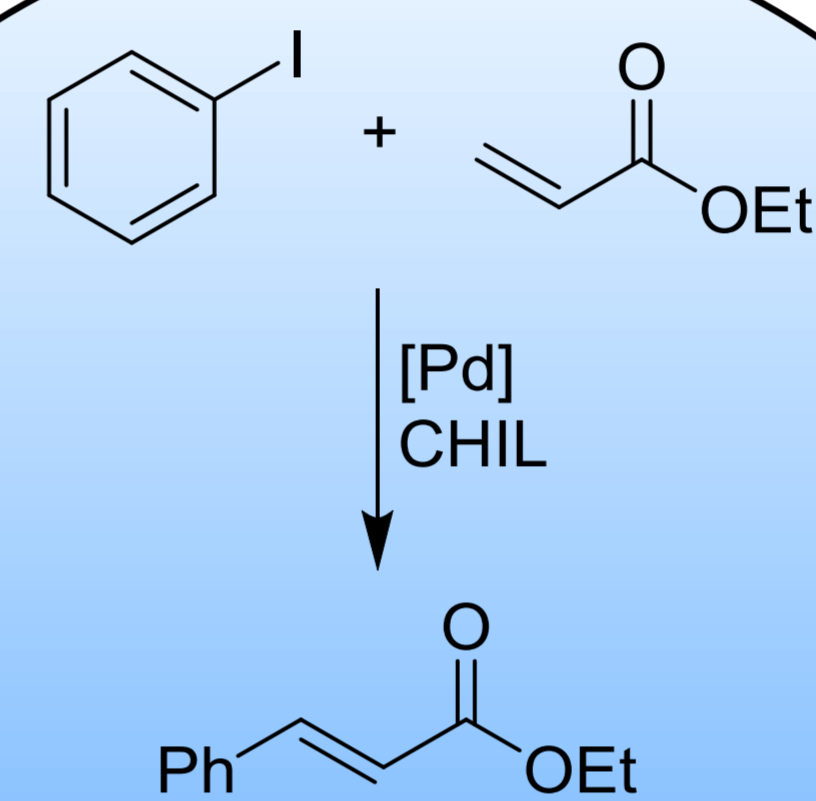
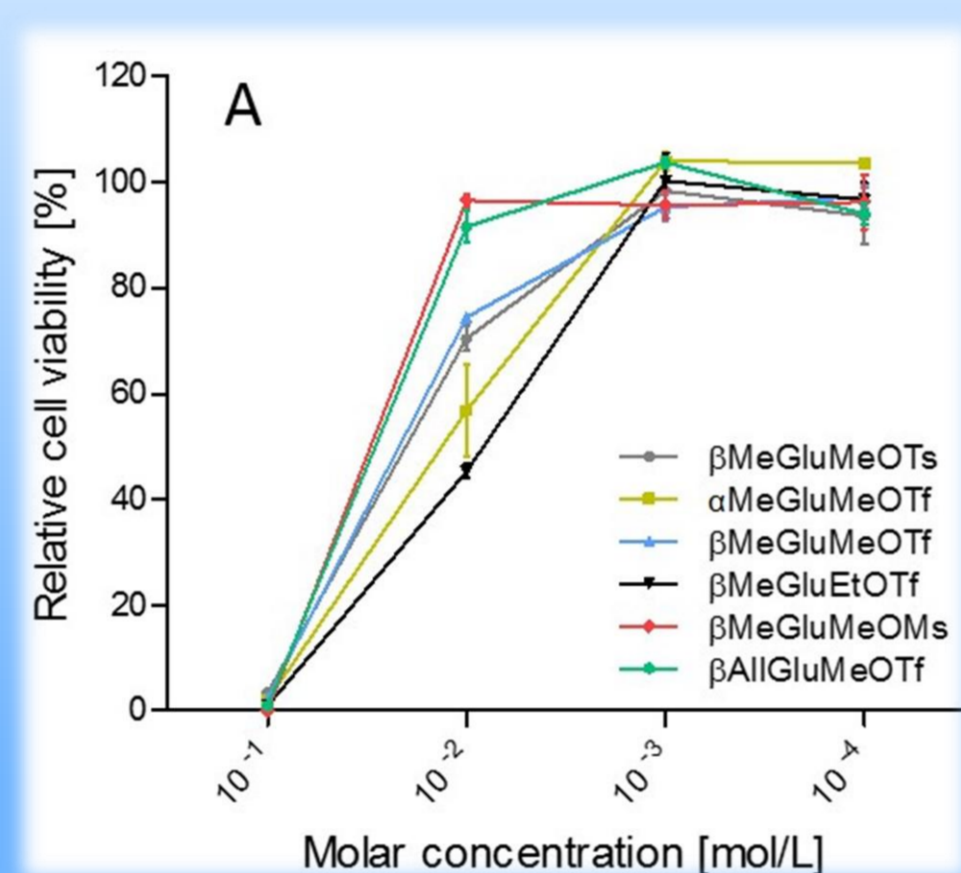


Junior Research Group Carbohydrate Chemistry - CHILs

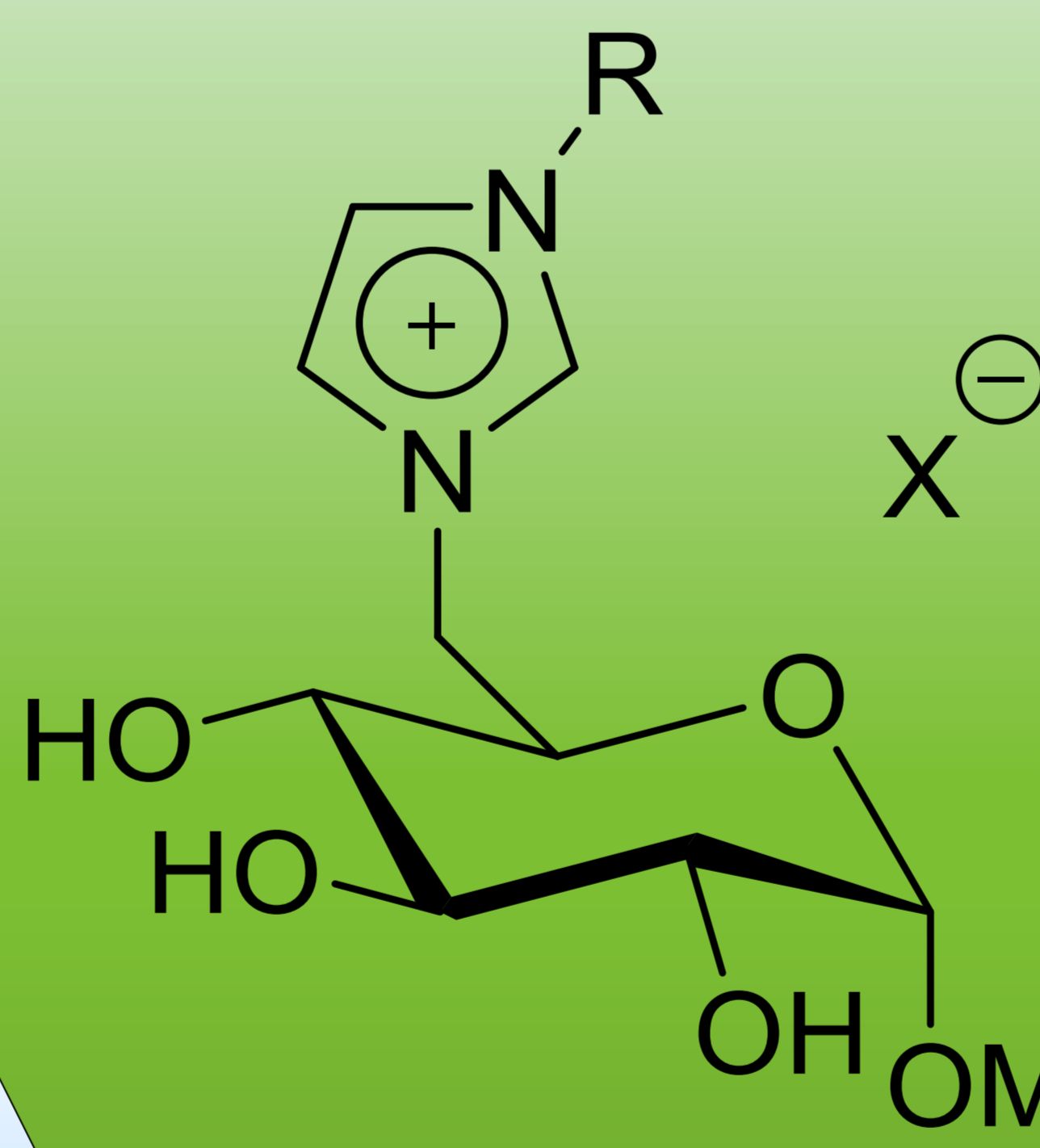
Ionic Liquids have become a widely known class of chemicals since their discovery. Defined as salts with melting points below 100°C, ionic liquids exhibit many useful properties like low vapor pressure, high thermal stability and exceptional dissolution ability. Commonly used cations for ILs are imidazolium, pyridinium or tetraalkylammonium based, while the anions are often weakly coordinating anions like BF_4^- , NTf_2^- or OTf^- . Many sub-classes of ILs have been developed since then. Some examples are the room temperature ILs (RTILs), which are useful as solvents, the polymerized ionic liquids (PILs), which function as charged polymers and the chiral ILs (CILs), which can be used in asymmetric syntheses. In the case of chiral ionic liquids, an effective strategy is to create ILs from the natural chiral pool, including amino acids, terpenes and carbohydrates. Our research group focuses on carbohydrate based chiral ionic liquids (CHILs) – a relatively unexplored class of ILs with great potential as green alternatives to common ionic liquids. Our goal is the development and fundamental understanding of new CHILs as well as to explore suitable applications.

Biocompatibility

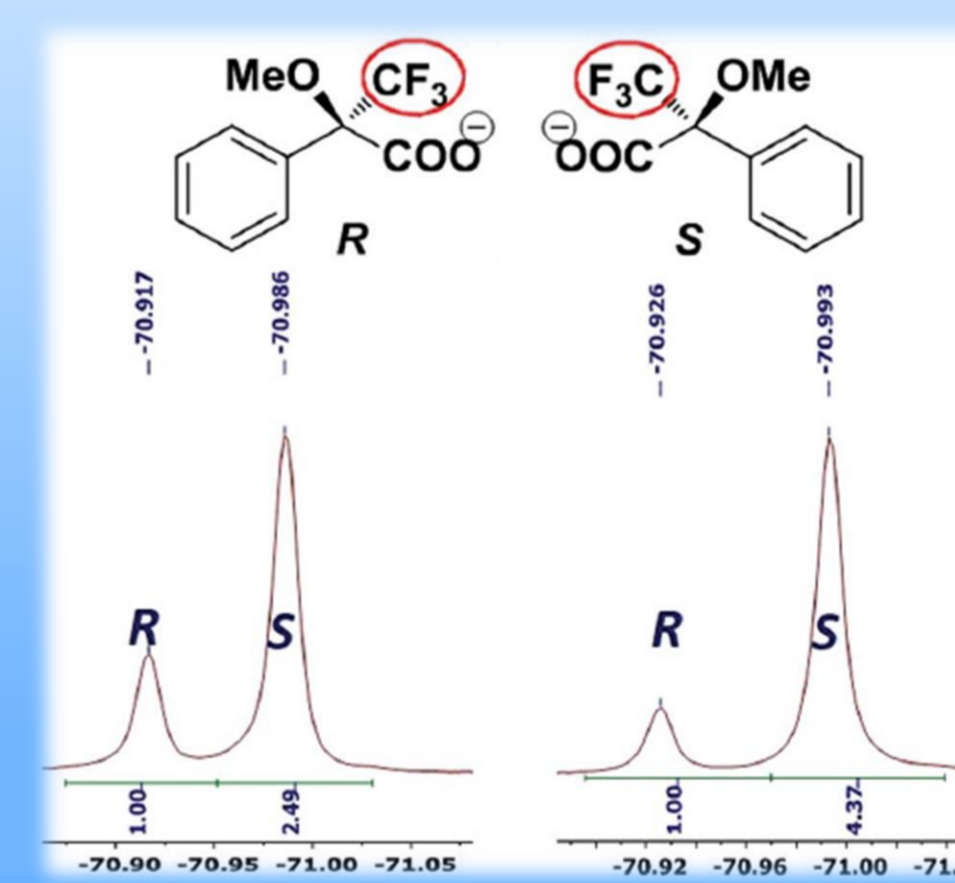


New Solvents in Organic Synthesis

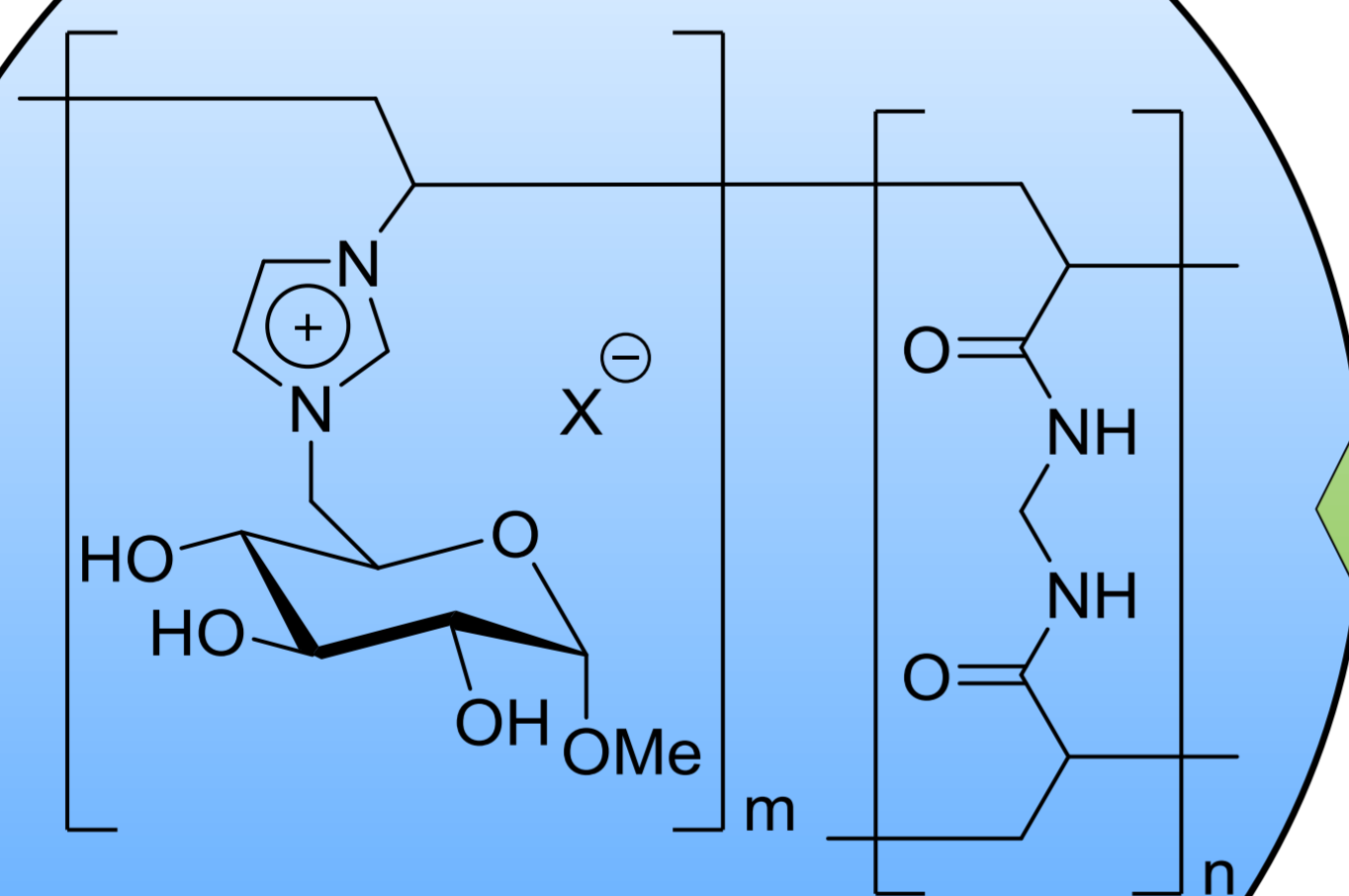
Carbohydrate based Ionic Liquids



Enantiomeric Recognition by NMR



Polymers



melting point:

$\text{NTf}_2^- < \text{OTf}^- < \text{OMs}^- < \text{I}^- < \text{OTs}^-$

decomposition point:

$\text{I}^- = \text{OMs}^- = \text{OTs}^- < \text{OTf}^- < \text{NTf}_2^-$

Structure Property Relationships

Cellulose Dissolution

