

Alcohol Dehydrogenase-catalyzed Enantioselective Synthesis of α -Halogenated Alcohols Bearing a Pyridine Ring

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Introduction

The substrate range of the (*R*)-selective alcohol dehydrogenase evo-1.1.270 towards prochiral α -halogenated ketones that are linked via a methylene group with a pyridine ring was investigated. In contrast to the intensively studied acetophenone and pyridine derivatives^[1] this kind of structural motive is rarely investigated for biocatalytic reduction reactions. High conversions with excellent enantioselectivities were found for relatively small substrates with two or less halogen substitutions. Substrates with even larger heterocycles and more halogen substitutions were not converted.

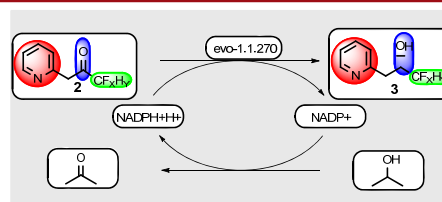


Figure 1. evo-1.1.270-catalyzed synthesis of enantiopure pyridinylpropanols

Results

The investigated ketones were synthesized starting from the respective picoline 1a-d or quinanzoline derivatives 1e using the lithium bases *n*-butyllithium (*n*BuLi) or lithium diisopropyl amide (LDA). The different α -halogenated acyl moieties were introduced through the use of different esters or the use of dimethylacetamide (DMA). The straight-forward one pot synthesis was followed by simple workup consisting of an aqueous extraction, while in some examples an additional column chromatography step was required.

Substrates 2a-c and 2e-g were converted with yields between 36% and 98% and very high values for the enantiomeric excesses of up to > 99%. The positive effect of α -halogenation (CFH₂ and CF₂H groups) is clearly shown by high equilibrium yields, even for complex substrates. The introduction of an CF₃ group or substitution on the heterocycle causes a full loss of reactivity (2d, 3h-3o). A similar behavior was also reported recently for α -halogenated acetophenone derivatives.^[2]

Table 1. Substrate range for subsequent enzymatic reduction

1,2 R ¹		R ²	1,2 R ¹		R ²
a	CH ₃	4-(NC ₅ H ₄)	j	CH ₃	2-(NC ₅ H ₃ CH ₃)
b	CH ₂ F	4-(NC ₅ H ₄)	k	CF ₃	2-(NC ₅ H ₃ CH ₃)
c	CHF ₂	4-(NC ₅ H ₄)	l	CH ₃	2-(NC ₅ H ₂ (CH ₃) ₂)
d	CF ₃	4-(NC ₅ H ₄)	m	CF ₃	2-(NC ₅ H ₂ (CH ₃) ₂)
e	CH ₃	2-(NC ₅ H ₄)	n	CH ₃	2-(NC ₉ H ₆)
f	CH ₂ F	2-(NC ₅ H ₄)	o	CF ₃	2-(NC ₉ H ₆)
g	CHF ₂	2-(NC ₅ H ₄)			
h	CF ₃	2-(NC ₅ H ₄)			
i	CClF ₂	2-(NC ₅ H ₄)			

The shown library of synthesized ketones (table 1) was converted to the corresponding alcohols with very high *ee* and good yields. The obtained results of the enzymatic reduction show that almost half of the investigated substrates were successfully transformed into chiral alcohols, while especially larger substrates were not accepted as substrates (table 2).

Table 2. Conversion of prochiral ketones 2a-2o to the corresponding chiral alcohols 3a-3o.

2,3	R ¹	R ²	Y [%]	<i>ee</i>
a	CH ₃	4-(NC ₅ H ₄)	93	>99(R)
b	CH ₂ F	4-(NC ₅ H ₄)	98	n.d.
c	CHF ₂	4-(NC ₅ H ₄)	95	>99(S)
d	CF ₃	4-(NC ₅ H ₄)	0	-
e	CH ₃	2-(NC ₅ H ₄)	36	>99(R)
f	CH ₂ F	2-(NC ₅ H ₄)	70	95(S)
g	CHF ₂	2-(NC ₅ H ₄)	60	n.d.
h	CF ₃	2-(NC ₅ H ₄)	0	-
i	CClF ₂	2-(NC ₅ H ₄)	0	-

Limitations:

- Only small heterocyclic moieties are accepted (2a-c; 2e-g).
- The reactivity increases with increasing number of substitutions.
CH₃ < CH₂X < CHX₂
- Full loss of reactivity was obtained with CX₃ substitution (2d; 2h-i).

Summary and Outlook

The aim of this study was to examine the reduction of α -halogenated ketones that are linked via a methylene group with a pyridine ring by a commercially available (*R*)-selective alcohol dehydrogenase. The investigation showed that α -halogenation has a positive effect on the equilibrium yield. Furthermore this study shows that large heterocyclic moieties as well as CF₃ substitution on the ketone

remain challenging substrates. Taking this limitations into account we found a series of CH₃, CH₂F and CHF₂ α -substituted pyridine derivatives that are easily converted from the methyl ketones to the corresponding chiral alcohol. The obtained alcohols exhibit high *ee* values of up to >99%.

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